
UNIVERSITI SAINS MALAYSIA

Peperiksaan Semester Kedua
Sidang Akademik 2005/2006

April/Mei 2006

EBB 202/3 - Kristalografi & Ikatan Dalam Pepejal

Masa : 3 jam

Sila pastikan bahawa kertas peperiksaan ini mengandungi LIMA muka surat beserta DUA muka surat (Lampiran) yang bercetak sebelum anda memulakan peperiksaan.

Kertas soalan ini mengandungi TUJUH soalan.

Kertas soalan ini mengandungi BAHAGIAN A, BAHAGIAN B dan BAHAGIAN C. Semua soalan di BAHAGIAN A adalah WAJIB dijawab. Calon hendaklah menjawab DUA soalan dari BAHAGIAN B dan DUA soalan dari BAHAGIAN C. Jawab LIMA soalan. Jika calon menjawab lebih daripada lima soalan hanya lima soalan pertama mengikut susunan dalam skrip jawapan akan diberi markah.

Jika calon menjawab SOALAN 2 dan 4, LAMPIRAN 1 dan 2 hendaklah disertakan bersama-sama dengan buku jawapan.

Mulakan jawapan anda untuk setiap soalan pada muka surat yang baru.

Semua soalan mesti dijawab dalam Bahasa Malaysia.

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BAHAGIAN A

1. [a] Terangkan secara ringkas mengenai yang berikut:
- (i) Model Atom
 - (ii) Kumpulan Simetri Titik
 - (iii) Kumpulan Simetri Ruang
- (50 markah)
- [b] Terangkan mekanisma penyerapan sinar-X dan tuliskan persamaan bagi keamatan alur sinar-X yang melalui satu bahan homogen berjarak x .
- (30 markah)
- [c] Sebanyak separuh keamatan sinar-X homogen telah ditapis oleh penapis aluminium berketebalan 10 mm. Kirakan pecahan sinaran ini jika 20 mm aluminium digunakan.
- (20 markah)

BAHAGIAN B

2. [a] Terangkan (beserta lakaran yang jelas) mengenai elemen-elemen simetri (termasuk simetri peralihan) yang terlibat dalam membangunkan kumpulan simetri titik dan simetri ruang.
(50 markah)
- [b] Lukis imej penuh bagi unjuran stereografik yang diberi dalam **Lampiran 1**.
(50 markah)
3. [a] Apakah kepentingan kumpulan simetri titik dan kumpulan simetri ruang dalam pencirian hablur? Jelaskan bagaimana **KEDUA-DUA** kumpulan simetri tersebut dapat dibangunkan.
(70 markah)
- [b] Berikan persamaan am gelombang Schrödinger dan apakah yang dimaksudkan dengan rangkap gelombang, ψ , dan ψ^2 . Apakah yang diberikan oleh penyelesaian persamaan ini?
(30 markah)
4. [a] Lukiskan atau tandakan elemen-elemen simetri yang dipunyai oleh hablur-hablur yang diberikan dalam **Lampiran 2**. Dalam kumpulan simetri titik manakah hablur-hablur tersebut tergolong?
(60 markah)
- [b] Setiap bahan hablur menunjukkan sifat-sifat yang berbeza di antara satu sama lain dan terdapat beberapa sifat bahan yang mempunyai kaitan langsung dengan jenis ikatan dalam hablur.
- (i) Jelaskan jenis-jenis ikatan pepejal.
 - (ii) Apakah sifat-sifat bahan yang berkait dengan jenis ikatan?
 - (iii) Bagaimanakah ikatan dalam pepejal mempengaruhi sifat-sifat tersebut?
- (40 markah)

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BAHAGIAN C

5. [a] Terangkan perbezaan mekanisma penghasilan sinar-X dan penghasilan kesan foton.

(80 markah)

- [b] Kirakan frekuensi ambang (λ_K) dalam unit Å bagi sinar-X yang dihasilkan oleh 50 keV elektron dalam tiub perlanggaran.

[Diberi $h = 6.63 \times 10^{-34}$ J.s, $c = 3 \times 10^8$ m/s, $e = 1.6 \times 10^{-19}$ J/eV]

(20 markah)

6. Plotkan graf keamatan (I/I_0) lawan sudut serakan (2θ) untuk Nikel (Ni) berdasarkan Fail Data Serbuk Sinar-X JCPDS (lihat Rajah 1).

(100 markah)

4-0850 MINOR CORRECTION

d	2.02	1.76	1.25	2.094	Ni					
4-0850										
I/I ₀	100	42	21	100	Nickel					
4-0850										
Rad. CuKα	λ 1.5405		Filter Ni		d Å	I/I ₀	hkl	d Å	I/I ₀	hkl
Dis.	Cut off		Coll.		2.094	100	111			
I/I ₀ G. C. DIFFRACTOMETER			d corr. abs.?		1.762	42	200			
Ref. SWANSON AND TATGE, JC FEL. REPORTS, 1951			REFS		1.248	21	220			
					1.0624	20	311			
Sys. Cubic			S.G. ON - F ₄₃₂		1.0172	7	222			
a ₀ 3.5238 Å	a ₀	Å	C		0.8810	4	400			
h	h	h	h		.8094	14	331			
Ref. 1812.	Y	Σ 4			.7880	15	420			
2θ	2θ	2θ	Sign							
2V	D ₅₀ 907 mp	Color								
Ref.										
SPECTROGRAPHIC ANALYSIS SHOWS <0.01% EACH OF Mn, Si AND Ca.										
AT 26°C										
TO REPLACE 1-1258, 1-1260, 1-1266, 1-1272,										
3-1043, 3-1051										

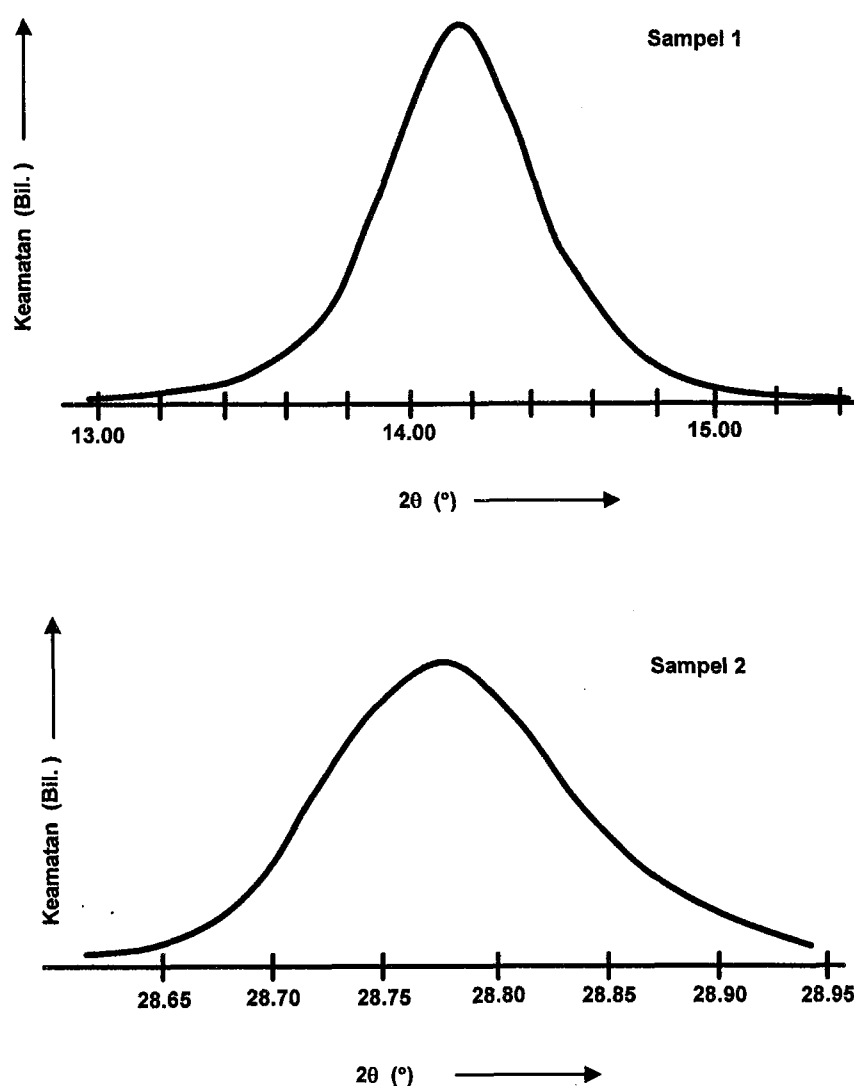
Rajah 1 - Fail Data Sinar-X JCPDS bagi Nikel.

7. [a] Terbitkan formula Scherrer.

(50 markah)

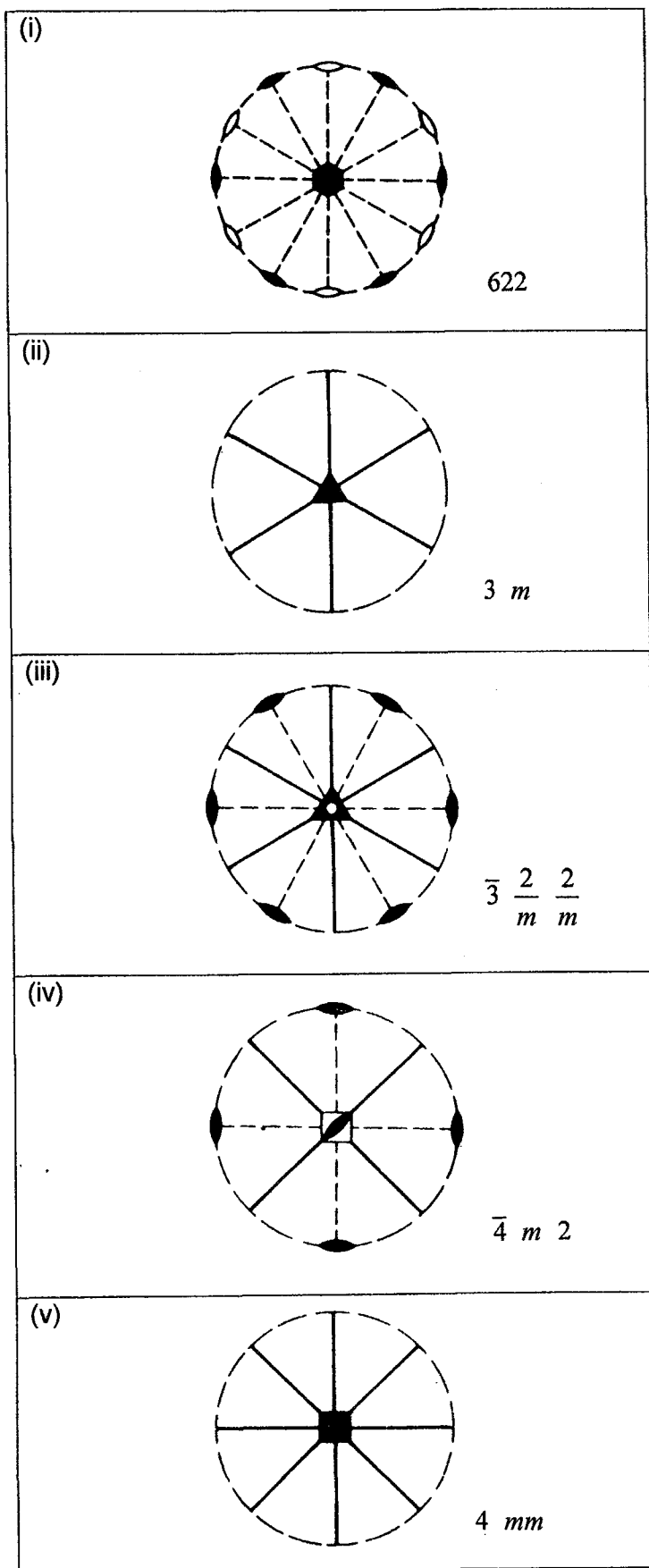
- [b] Rajah 2 menunjukkan dua puncak XRD bagi polimer-garam terkompleks. Kirakan panjang Scherrer dan tentukan sampel yang mana mempunyai darjah kristal yang tinggi. [Diberi $\lambda = 1.5418\text{\AA}$]

(50 markah)

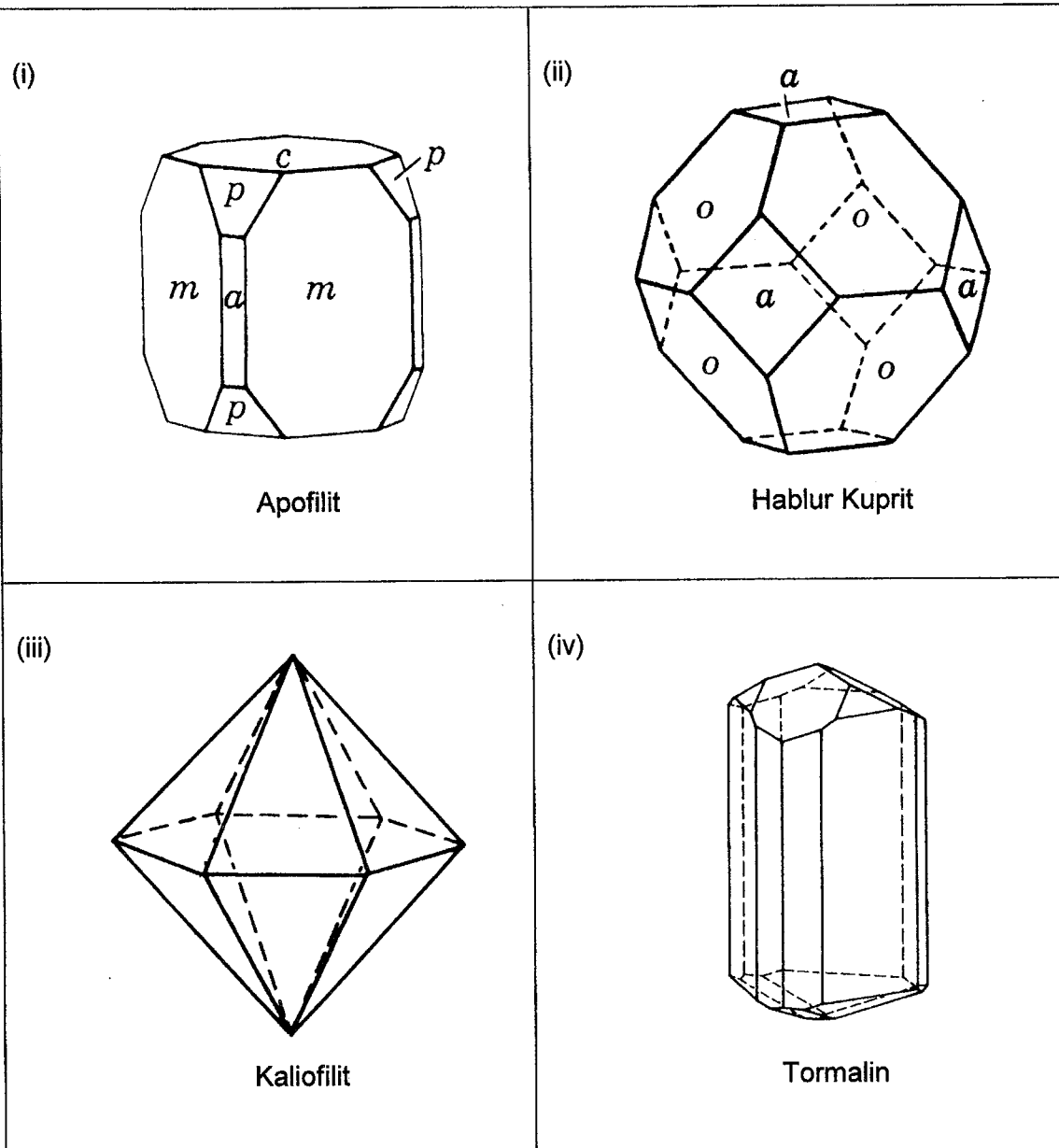


Rajah 2 - Puncak-puncak XRD bagi polimer-garam terkompleks

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LAMPIRAN 1

LAMPIRAN 2



TRANSLATION

UNIVERSITI SAINS MALAYSIA

Second Semester Examination
Academic Session of 2005/2006

April/May 2006

EBB 202/3 - Crystallography & Bonding in Solids

Time : 3 hours

Please check that this examination paper consists of FIVE pages of printed material and TWO pages APPENDIX before you begin the examination.

This paper contains SEVEN questions.

This paper contains SECTION A, SECTION B and SECTION C. Answer all questions from SECTION A. Answer TWO questions from SECTION B and TWO questions from SECTION C. Answer any FIVE questions. If a candidate answers more than five questions, only the first five answered will be examined and awarded marks.

If you answer QUESTION 2 and 4, APPENDIX 1 and 2 must be attached together with the answer script.

Answer to any question must start on a new page.

All questions must be answered in Bahasa Malaysia.

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PART A

1. [a] Explain briefly about the following:
- (i) Atomic Model
 - (ii) Point Group Symmetry
 - (iii) Space Group Symmetry
- (50 marks)
- [b] Explain the mechanisms of X-rays absorption and write down an expression for the intensity of X-ray beam as it passes through x distance any homogeneous substance.
- (30 marks)
- [c] One-half the intensity of a homogeneous X-ray beam is removed by aluminium filter 10 mm thick. Calculate the fraction of this homogeneous beam that would be removed by 20 mm of aluminium.
- (20 marks)

PART B

2. [a] Explain (with clear sketch/drawing) about the symmetry elements (including translational symmetry) involved in developing point group symmetry and space group symmetry.
(50 marks)
- [b] Draw the full image for the stereographic projection given in **Appendix 1**.
(30 marks)
3. [a] What is the importance of point group symmetry and space group symmetry in the characterization of crystals? Explain how **BOTH** symmetry groups can be developed.
(70 marks)
- [b] Give the Schrödinger wave equation and what is meant by wave function ψ and ψ^2 . What does the solved equation gives?
(30 marks)
4. [a] Draw or labeled the symmetry elements that all the crystals given in **Appendix 2** possessed. In which point group symmetry does the crystals belong to?
(60 marks)
- [b] All crystal materials showed different properties from one to the other and there are a few properties which directly related to the type of bonding in crystal.
- (i) Explain the types of bonding in solid.
 - (ii) What are the properties related to types of bonding?
 - (iii) How does the bonding in solid influence those properties?
- (40 marks)

PART C

5. [a] Explain the difference in the mechanism to produces X-rays and photon effect.

(80 marks)

- [b] Determine the cut-off wavelength (λ_K) in unit Å of x-rays produced by 50 keV electron in a Coolidge tube.

[Given $h = 6.63 \times 10^{-34}$ J.s, $c = 3 \times 10^8$ m/s, $e = 1.6 \times 10^{-19}$ J/eV]

(20 marks)

6. Plot a graph intensity (I/I_1) versus 2θ for Nickel (Ni) based on JCPDS X-Ray Powder Data File (see Figure 1). [Given $\lambda = 1.5418\text{Å}$]

(100 marks)

4-0850 MINOR CORRECTION

d	2.03	1.76	1.25	2.094	Ni
4-0854					
$1/\lambda$	100	42	21	100	NICKEL
4-0850					
Rad. CuK α	λ 1.5405	Filter Ni	d Å	$1/\lambda$	hkl
Di. Cut off	Coll.		2.094	100	111
$1/\lambda$ G. C. DIFFRACTOMETER	d corr. abs.?		1.762	42	200
Ref. SWANSON AND TATGE, J.C. FEL. REPORTS, 1951			1.246	21	220
			1.0624	20	311
			1.0172	7	222
Sys. CUBIC	S.G. $O_h^5 - Fm3m$		0.8810	4	400
a 2.5238 Å	a Å	A C	.8084	14	331
b	b Å	Z 4	.7880	15	420
c	c Å				
Ref. Ibid.					
ρ 8.907 g/cm ³	ρ g/cm ³	ρ g/cm ³			
Ref.					
SPECTROGRAPHIC ANALYSIS SHOWS 40.01% EACH OF Fe, Si AND Ca.					
AT 26°C					
TO REPLACE 1-1258, 1-1260, 1-1266, 1-1272, 3-1043, 3-1051					

Figure 1 - Nickel JCPDS X-Ray Powder Data File

7. [a] Derive the Scherrer's formula.

(50 marks)

- [b] Figure 2 shows the XRD peaks of two polymer-salt complexed samples. Calculate the Scherrer length and determine which sample has higher degree of crystallinity. [Given $\lambda = 1.5418\text{\AA}$]

(50 marks)

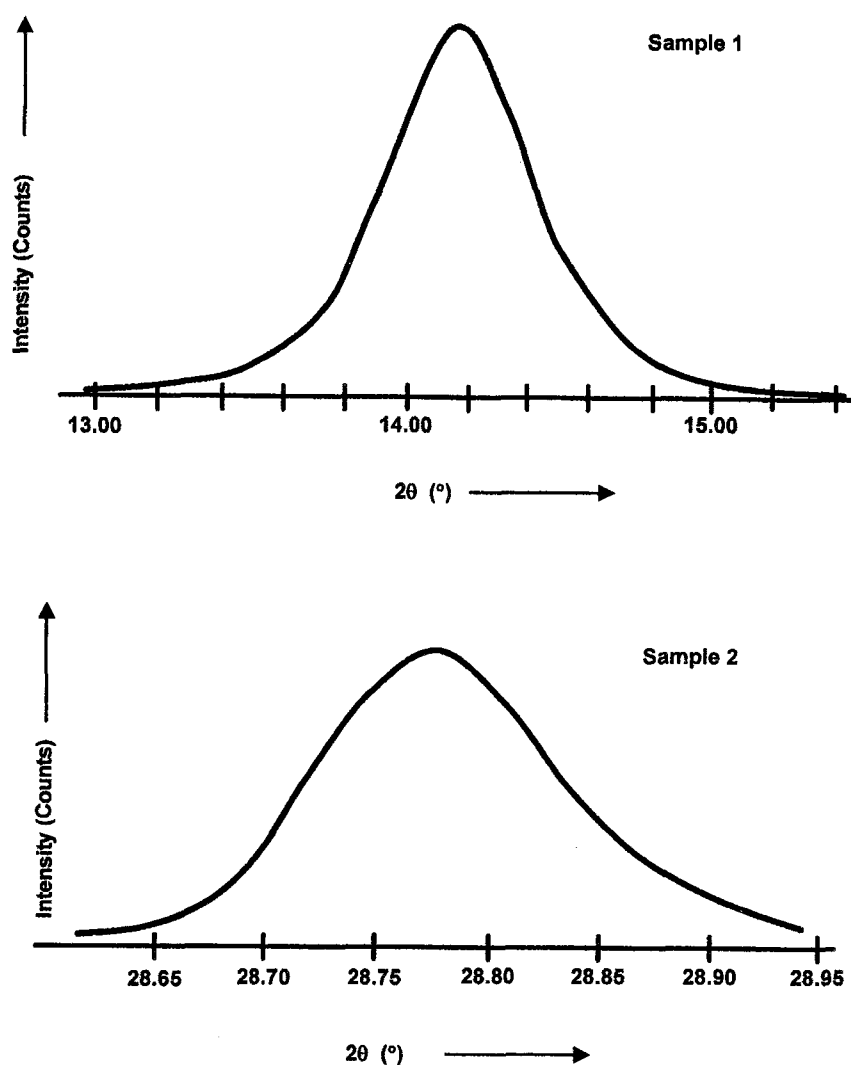


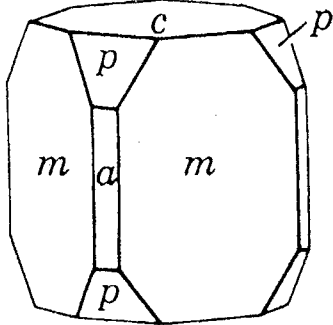
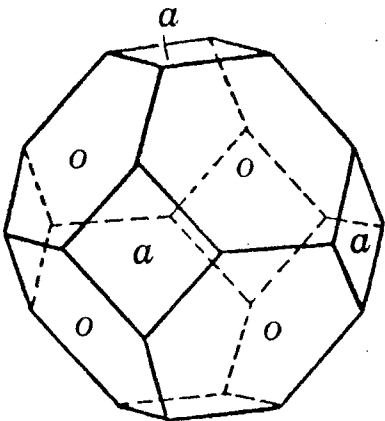
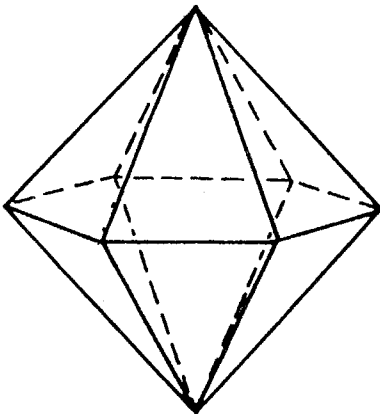
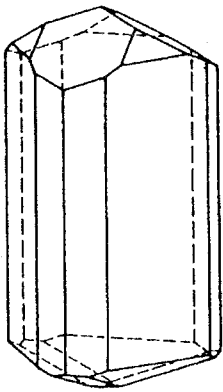
Figure 2 - XRD peaks for polymer-salt complexed

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APPENDIX 1

(i)		622
(ii)		3 <i>m</i>
(iii)		$\bar{3} \frac{2}{m} \frac{2}{m}$
(iv)		$\bar{4} m 2$
(v)		4 <i>mm</i>

APPENDIX 2

<p>(i)</p>  <p>Diagram of Apophyllite crystal habit. The crystal is a rounded, elongated form with a central vertical prism labeled 'a'. The prism is flanked by two large, rounded faces labeled 'm'. The top and bottom of the crystal are labeled 'c' and 'p' respectively. The sides are labeled 'p'.</p> <p>Apophyllite</p>	<p>(ii)</p>  <p>Diagram of Cuprite Crystals. The crystal is a rounded, elongated form with a central vertical prism labeled 'a'. The prism is flanked by two large, rounded faces labeled 'o'. The top and bottom of the crystal are labeled 'a' and 'o' respectively. The sides are labeled 'o'.</p> <p>Cuprite Crystals</p>
<p>(iii)</p>  <p>Diagram of Kaliophilite crystal habit. The crystal is a tetrahedral form with a central vertical prism labeled 'a'. The prism is flanked by two large, rounded faces labeled 'm'. The top and bottom of the crystal are labeled 'c' and 'p' respectively. The sides are labeled 'p'.</p> <p>Kaliophilite</p>	<p>(iv)</p>  <p>Diagram of Tourmaline crystal habit. The crystal is a prismatic form with a central vertical prism labeled 'a'. The prism is flanked by two large, rounded faces labeled 'o'. The top and bottom of the crystal are labeled 'a' and 'o' respectively. The sides are labeled 'o'.</p> <p>Tourmaline</p>